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~~Introduction to acids
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19: Acids and Bases
By: Loreley Estrada
Stephanie Sipprelle~~

Acids and Bases

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~~Acids Bases and~~

~~Salts *Make Your Own*~~

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~~Acids, Bases, and pH~~

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Naming Acids

Introduction

BOOK 2 CHAPTER 5

CONCEPT

*TRAILER? (Piggy
Predictions)*

*Acids,
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11 chapter 7 |

Equilibrium | Ionic

Equilibrium 01 |

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STUDY. PLAY.

Characteristics of
Acids. they taste sour,
will change the color
of an acid-base
indicator, and can be
strong or weak

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electrolytes in an
aqueous solution.

Characteristics of
Bases.

Answers

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Jennie L. Borders.

Section 19.1 – Acid-
Base Theories. Acids
have a sour taste,
change the color of an
indicator, can be
strong or weak

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electrolytes in
aqueous solution, and
react with metals.
Bases.

Answers

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and Bases.

Description.

Acids/Bases

descriptions &
formulas with pH.

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Bases, and Salts

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Brønsted-Lowry acids
and bases come in

pairs. A conjugate base is the remainder of the B-L acid, after it donates its H^+ . A conjugate acid is the substance formed when the B-L base gains a H^+ . Thus, a conjugate acid-base pair is related by the

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Acids And Bases

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Answers

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Neutralization

Reactions > Chapter
19 Acids, Bases, and
Salts 19.1 Acid-Base
Theories 19.2

Hydrogen Ions and
Acidity 19.3 Strengths

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of Acids and Bases

19.4
And Salts

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Neutralization

Reactions > Chapter
19 ...

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Bases, and Salts.

STUDY. PLAY.

Tastes sour. Acid.

Changes the color of
an acid-base indicator
(acid, base, or both)

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Both. Can be strong or weak electrolytes in aqueous solution.

Acid. In drinks, citrus, pop, etc. Acid. This acid is associated with protein. amino acid. Used in fragrances and flavors.

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Bases, and Salts
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Chemistry Chapter
19: Bases and Acids.
STUDY. PLAY. Acids.
taste sour, change the
color of an acid base
indicator, can be
strong or weak
electrolytes in
aqueous solution Ex:
citrus. bases. taste
bitter, feel slippery,
will change the color
of an acid base
indicator, can be

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strong or weak
electrolytes in aqueous
solution. Ex: soap

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Chemistry - Chp 19 -
Acids, Bases, and
Salt - PowerPoint
Chapter 19 Acids and
Bases Now, consider
the equation for the
ionization of hydrogen

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fluoride in water.

According to the Brønsted-Lowry definition, the

equation is written this way. Agricultural

Technician

$\text{HF} + 3$

H_2O

$\text{H}_3\text{O}^+ + \text{F}^-$

What are the conjugate acid-base pairs?

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Chap19.pdf -

CHAPTER 19 Acids
and Bases What
You\u2019ll ...

Chapter 19 Acids,
Bases, and Salts

?What are the
properties of acids
and bases? acids
taste sour, will change
the color of an acid-
base indicator, and
can be strong or weak
electrolytes in

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Chapter 19 Acids,
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Acids and Bases:
Basics. These are
really just a specific
type of chemical
compound. No

mysteries here. 1)

They **MUST** be ionic

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compounds, and

MUST dissolve in water, that's the only way to be chemically active. We call this.

disassociation.

Chapter 19 Acids, Bases, and Salts
of an acid or base is dissolved in solution - refers to the # of moles of acid or base.
Is a . concentrated,

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weak. acid possible?

Acid and Base

Strength. Acetate is a
stronger base than H.

2. O, LOWER on

table, The stronger
base “wins” the

proton. ... Chapter 19

Acids, Bases, and
Salts

Chapter 19 Acids,
Bases, and Salts -
MOLEBUS

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(ALLCHEM)

conjugate acid-base
pair amphoteric

Section 19.1 Acids

and Bases: An

Introduction When

ants sense danger to

the ant colony, they

emit a substance

called formic acid that

alerts the entire

colony. Acids in

rainwater hollow out

enormous limestone

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caverns and destroy
valuable buildings and
statues. Acids flavor

Chapter 19: Acids and Bases

Chapter 19: Acids and
Bases You're
probably already
aware from what
you've learned in
other classes that
acids and bases are
all over the place.

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From the lemon you put in iced tea to the Drano put down the drain to keep your hair from clogging the drain, acids and bases are an important part of our lives.

Chapter 19: Acids and
Bases

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Winston Churchill

High. Acids, Bases,
and Salts Acid-Base
Theories

OBJECTIVES: Define
the properties of acids
and bases. Acid-Base
Theories

OBJECTIVES:
Compare and

Chapter_19 - Acids
Bases and Salts Acid-
Base Theories ...

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Acids and Bases.

SC6. Obtain, evaluate, and communicate information about the properties that describe solutions and the nature of acids and bases. f. Use mathematics and computational thinking to compare, contrast, and evaluate the nature of acids

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and bases in terms of
percent dissociation,
hydronium ion
concentration, and
pH. g.

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