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an single atom that has gained or lost one or more electrons
a.... a positively charged ion. a negatively charged ion. States
that atoms lose, gain or share electrons in order to ac....

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Periodic Law SECTION 1 SHORT ANSWER Answer the
following questions in the space provided. 1. c In the modern
periodic table, elements are ordered (a) according to
decreasing atomic mass. (b) according to Mendeleev's
original design. Modern ...

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Chemistry Chapter 5 Exam. Multiple Choice. Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a.

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Chemistry: The Central Science (13th Edition) Brown, Theodore E.; LeMay, H. Eugene; Bursten, Bruce E.; Murphy, Catherine; Woodward, Patrick; Stoltzfus, Matthew E. Publisher. Prentice Hall. ISBN.

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Kerala State Syllabus 10th Standard Chemistry Solutions Chapter 5 Compounds of Non-Metals Compounds of Non-Metals Text Book Questions and Answers. Text Book Page No: 79. SSLC Chemistry Chapter 5 Question 1. Take a little ammonium chloride (NH_4Cl) in a watch glass and add a little calcium hydroxide ($\text{Ca}(\text{OH})_2$) to it. Stir well. Can you sense any smell?

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Answer: In heterogeneous catalysis, the reactants are generally gases whereas catalyst is a solid. The reactant molecules are adsorbed on the surface of the solid catalyst. As a result, the concentration of the reactant molecules on the surface increases and hence the rate of reaction increases. Question 8.

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(a) The enzyme activity is maximum at optimum pH which is between 5-7 (b) Each enzyme is specific for a given reaction (c) The favourable temperature range of enzyme activity is between 25-37°C (d) The enzymatic activity is increased in presence of certain substances called co-enzymes. Answer

~~MCQ Questions for Class 12 Chemistry Chapter 5 Surface ...~~

Modern Chemistry 37 The Periodic Law CHAPTER 5

REVIEW The Periodic Law SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1.

_____ When an electron is added to a neutral atom, energy is (a) always absorbed.

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an single atom that has gained or lost one or more electrons
a.... a positively charged ion. a negatively charged ion. States
that atoms lose, gain or share electrons in order to ac....
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