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34 ChemQuest 11 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Energy Levels and Sublevels As you know, in his solar system model Bohr proposed that electrons are located in energy levels. The current model of the atom isn't as simple as that, however.

~~34 ChemQuest 11 Name: Yumpu~~

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JI ChemQuest 34 Name: Date Hour::F-F Fluorine Gas Molar mass = 38.0 g  
c.t- Q1 Chlorine Gas Molar mass ::Br BI Bromine Liquid Molar mass : I-  
I' Iodine Solid Molar mass : Information: Some Molecules and Their  
States Critical Thinking Questions What type of force exists between  
two Fz molecules--{ispersion, dipolar, or hydrogen bonds?,".eh"s#n  
(€l'5Feq-ece\$'1 What t5.pe of force exists between ...

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the same exponents. Your answer will then have the same exponent as  
the numbers you add or subtract. Here are some examples. Make sure  
you understand each step:  $4.2 \times 10^6 + 3.1 \times 10^5$  make exponents the same,  
either a 5 or 6  $4.2 \times 10^5 + 3.1 \times 10^5 = 4.51 \times 10^5 = 4.51 \times 10^6$

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ChemActivity 50: Electrochemical Cell UNIT 11 - HW Practice Keys -  
ChemQuest 55: Free Energy - ChemQuest 54: 2nd Law of Thermodynamics

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32 Information: Light Recall that light is a wave. White light is composed of all the colors of light in the rainbow. All light travels

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at the same speed ( $c$ ),  $3.00 \times 10^8$  m/s. (The speed of all light in a vacuum is always equal to

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO<sub>2</sub> are produced, so the number of moles of CO<sub>2</sub> that can be produced when 4 moles of propane combusts =  $4(3) = 12$ . ... ChemQuest 34. Name: \_\_\_\_\_ Date ...

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A certain element exists as three different isotopes. 24.1% of all isotopes have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu.

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